

Extraction of carotenoids from crustacean waste using organic solvents

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ABSTRACT

Carotenoid pigments are divided into two groups, xanthophylls (which contain oxygen) and carotenes (which are pure hydrocarbons, and contain no oxygen) having a strong antioxidant ability. Thus, they are applicable in food, pharmaceutical and cosmetic industries. It has been reported that astaxanthin has up to 10 times antioxidant activity than other carotenoids such as zeaxanthin, lutein, canthaxanthin and β -carotene; and 100 times more than that of α -tocopherol. For carotenoid extraction, polar solvents such as isopropyl alcohol (IPA), acetone and etc., non-polar solvents such as petroleum ether, hexane and etc. or the mixture of them are used.

Keywords: carotenoids, crustacean waste, organic solvents, extraction

INTRODUCTION

Animals are not able to produce carotenoids, so they have to get it from food chain with feed from algae and plants. These pigments are found in aquatic animals such as salmon, birds such as flamingo, in microorganisms and algae. The carotenoid pigments are the main and important pigment in nature; more than 600 kinds of carotenoids are available. The aquatic animals such as crustacean are known to contain various carotenoids and are considered as one of the important sources of natural carotenoids. Astaxanthin and its esters are the main pigment in crustacean. Astaxanthin is in the group of xanthophylls that have an O₂ or OH instead of H in carotene. Astaxanthin is the main component that makes an orange – pink color in salmon and crustacean. These waste products are economically recoverable, because they have high quality protein, chitin, minerals, carotenoids, such as astaxanthin, and lipids that are high in ω -3 fatty acids. It has been reported that astaxanthin has up to 10 times the antioxidant activity than other carotenoids such as zeaxanthin, lutein, canthaxanthin and β -carotene; and 100 times more than that of α -tocopherol. In general, carotenoids and lipids are soluble in non-polar solvents and these types of solvent have

traditionally been used to extract them. Thus, several organic solvents have been permitted for use in food industries such as acetone, ethyl acetate, hexane, isopropanol, methanol, methyl ethyl ketone and ethanol. Other normally used solvents like dichloromethane, dimethyl sulfoxide and chloroform are not allowed, because of their toxicity. However, the disadvantage is that this method of extraction generally decreases the stability of carotenoids due to oxidation, if antioxidants are not used during and after processing. Environmental issues represent another concern (large volumes of solvent are required) and also the low extraction yields achieved using these types of solvent (almost 50% of the carotenoids are lost), stimulating studies directed at finding alternative extraction processes for this functional food and pharmaceutical supplement.

Sachindra *et al.* (2005) used organic solvents for extraction carotenoids from shrimp waste. In this research, polar, non-polar and the mixture of solvents were used. Hexane and isopropyl alcohol was the best solvent (Table 1).

Table 1. Yield of carotenoids from shrimp waste in different solvents and solvent mixtures (Sachindra *et al.*, 2005)

Solvent/solvent mixture	Yield ($\mu\text{g/g}$ waste) (WWB)*
Acetone	$40.6 \pm 1.6^{\text{a}}$
Methanol	$29.0 \pm 3.3^{\text{b}}$
Ethyl methyl ketone	$36.8 \pm 1.9^{\text{c}}$
Isopropyl alcohol (IPA)	$40.8 \pm 3.0^{\text{a}}$
Ethyl acetate	$36.9 \pm 2.9^{\text{c}}$
Ethanol	$31.9 \pm 2.2^{\text{d}}$
Petroleum ether	$12.1 \pm 1.8^{\text{e}}$
Hexane	$13.1 \pm 0.9^{\text{e}}$
Acetone:hexane (50:50)	$38.5 \pm 1.0^{\text{ac}}$
IPA:hexane (50:50)	$43.9 \pm 0.7^{\text{f}}$

* WWB – wet weight basis; values are means \pm SD ($n = 6$); values with different superscript (a–f) differ significantly ($p \leq 0.05$).

Khanafari *et al.* (2007) used chemical and microbial methods for extraction astaxanthin from shrimp waste. In this research, they used hexane for extraction of astaxanthin.

Sachindra *et al.* (2005) used acetone to extract carotenoids from marine and fresh waters crabs of Indian (Table 2, 3). They extracted carotenoids in different body components of Indian shrimps (Table 5). Sachindra *et al.* (2006) extracted carotenoids in *Solonocera indica* and *Aristeus alcocki* (Table 4).

**Table 2. Yield and total carotenoid content in meat and shell of crabs
(Sachindra *et al.*, 2005)**

	Marine crab (<i>C. cruciata</i>)		Fresh water crab (<i>P. potamon</i>)	
	Meat	Shell	Meat	Shell
Yield (g/100 g) ^a	29.7±3.9	34.4±1.2	28.8±1.5	35.7±1.0
Total carotenoid content (µg/g)(wwb) ^b	3.4±0.6	11.0±0.5	4.1±0.4	6.9±0.6

Values are mean±S.D (*n* = 6).

P ≤ 0.05 in meat between species.

P ≤ 0.001 in shell between species.

^a *P* ≥ 0.05 between species.

^b *P* ≤ 0.001 between meat and shell within species.

**Table 3. Composition (g/100 g of total carotenoids) of major carotenoids in
the carotenoid extract from marine and fresh water crab (by HPLC)
(Sachindra *et al.*, 2005)**

	Marine crab		Fresh water crab	
	Meat	Shell	Meat	Shell
Astaxanthin	17.3±1.1	23.6±1.4	9.3±0.6	7.2±1.0
Astaxanthin monoester	26.4±2.0	15.2±1.3	11.2±1.0	3.7±1.2
Astaxanthin diester	23.9±2.1	26.7±2.8	16.0±1.1	3.8±1.1
β-Carotene	3.6±1.3	5.1±1.4	7.4±1.6	3.6±1.2
Zeaxanthin	0.49±0.35	5.0±1.7	42.0±2.6	74.8±4.2
Unidentified	27.6±3.9	24.4±2.8	14.1±1.1	6.9±2.1

Values are mean±SD (*n* = 3).

**Table 4. Total carotenoid content (µg/g) in two species of deep-sea shrimp
(Sachindra *et al.*, 2006)**

Species	Body Component		
	Meat	Head	Carapace
<i>S. indica</i>	15.9 ± 2.1 ^{xa}	67.7 ± 6.3 ^{ya}	116.0 ± 11.8 ^{za}
<i>A. alcocki</i>	21.4 ± 1.7 ^{xb}	185.3 ± 17.0 ^{yb}	117.4 ± 6.7 ^{za}

Values are mean ± SD (*n* = 4); Different superscripts on values in individual columns (a, b) and rows (x, y, z) indicates significant difference (*p* ≤ 0.05)

Table 5. Total carotenoid content ($\mu\text{g g}^{-1}$) in different species of shrimp (Sachindra *et al.*, 2005)

Species	Meat	Head	Carapace
<i>Penaeus monodon</i>	17.4 ± 5.90 _{xa}	58.4 ± 7.73 _{yab}	86.6 ± 13.88 _{za}
<i>Penaeus indicus</i>	10.4 ± 0.92 _{xb}	35.8 ± 6.83 _{yb}	59.8 ± 11.02 _{zb}
<i>Metapenaeus dobsonii</i>	11.1 ± 1.61 _{xb}	51.3 ± 4.09 _{yb}	83.3 ± 13.87 _{za}
<i>Parapenaeopsis stylifera</i>	16.0 ± 2.21 _{xa}	153.1 ± 40.06 _{yc}	104.7 ± 11.39 _{zc}

Values are mean ± SD (n = 6).

a-c (columns), x-z (rows): significant difference ($p \leq 0.05$).

Coral-Hinostroza and Bjerkgeng (2002) used acetone:methanol for extraction of astaxanthin from red crab langostilla (*Pleuroncodes planipes*).

CONCLUSIONS

Iran has proximately 2400 km shoreline which makes it rich in crustacean such as crabs and shrimps. Iran fisheries organization announced that 1300 tons of shrimp fishing in 2000-2009 years. It shows that Iran have a strong potential in seafood industry. The crustacean wastes are rich in protein, carotenoid, chitin and etc. that can be recycled and make economical improvement in fisheries industry. In order to extract carotenoids from crustacean wastes, polar solvents such as isopropyl alcohol and acetone are used, however non-polar solvents must be used for the phase separation, or instead the mixture of polar and non-polar solvents can be used.

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